Two-band conduction and nesting instabilities in superconducting $Ba_2CuO_{3+\delta}$: First-principles study

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(Received 15 April 2021; revised 13 August 2021; accepted 16 August 2021; published 25 August 2021)

First-principles investigations of the high-temperature superconducting system $Ba_2CuO_{3+\delta_1}$ recently discovered at $\delta \approx 0.2$ at $T_c = 70$ K, are applied to demonstrate the effects of oxygen ordering on the electronic and magnetic properties. The observed "highly overdoped" superconducting phase displays stretched Cu-planar oxygen O_P distances and anomalously shortened Cu-apical O_A separations compared with other cuprates. The stoichiometric system $\delta = 0$, with its strongly one-dimensional (1D) Cu-O_P chain structure, when nonmagnetic shows 1D Fermi surfaces that lead, within density functional theory, to antiferromagnetic $Cu-O_P$ chains (a spin-Peierls instability). Accounting for 1D fluctuations and small interchain coupling according to the theory of Schulz indicates this system, like Sr₂CuO₃, is near the 1D Luttinger-liquid quantum critical phase. The unusual Cu-O bond lengths *per se* have limited effects on other properties for $\delta = 0$. We find that a "doubled bilayer" structure of alternating Cu-O_P chains and wide rung Cu₃O₄ ladders is the energetically preferred one of three possibilities where the additional oxygen ions bridge Cu-O_P chains in the superconducting phase $\delta = 1/4$. Nominal formal valences of the three Cu sites are discussed. The sixfold (octahedral) site is the most highly oxidized, accepting somewhat more holes in the d_{z^2} orbital than in the $d_{x^2-y^2}$ orbital. The implication is that two-band physics is involved in the pairing mechanism and the superconducting carriers. The Fermi surfaces of this metallic bilayer structure show both 1D and 2D strong (incipient) nesting instabilities, possibly accounting for the lack of clean single-phase samples based on this structure and suggesting importance for the pairing mechanism.

DOI: 10.1103/PhysRevB.104.054516

I. INTRODUCTION

In superconducting cuprates, the CuO₂ square lattice has been widely considered a crucial ingredient in producing high T_c due to the profuse number of examples [1]. However, some more structurally complex cuprates include Cu-O chains. An example of one class is the "telephone number compounds" exemplified by Sr₁₄Cu₂₄O₄₁ with a mean copper valence of Cu^{2.25+}, which is in the same doping regime as the superconducting square-lattice cuprates. For issues and references, see, e.g., Schmidt and Uhrig [2] and Armstrong *et al.* [3]. The Cu-O chains in these compounds were found to be plagued (or, depending on viewpoint, blessed) with inhomogeneities, likely related to one-dimensionality and spin dimerization. What is necessary, versus what is peripheral, for high- T_c superconductivity in cuprates remains under discussion.

Over 30 years ago [3–7] Ba₂CuO_{3+ δ} was synthesized in the range of $\delta = 0$ –0.4 and refined structurally based on the K₂NiF₄ structure of the $\delta = 1$ phase, i.e., a Ba₂CuO_{4- α} picture. At that time no superconductivity in the system was observed. Recently, Li *et al.* successfully obtained a bulk superconducting sample with a high critical temperature $T_c \approx 70$ K at $\delta \approx 0.2$ [8,9]. Their synthesis technique used a very high pressure of 18 GPa and high temperature of 1000 °C. Through x-ray absorption spectroscopy and resonant inelastic x-ray scattering measurements on a superconducting powder sample [10], Fumagalli *et al.* inferred two inequivalent Cu ions and an in-plane nearest neighbor superexchange $J \sim$ 0.12–0.18 eV. Theoretical studies [11–13] based on two-band models proposed unconventional superconductivity of either *d*-wave or s_{\pm} -wave pairing.

Compared with other superconducting cuprates, this extraordinarily heavily doped superconductor (Cu^{*p*}, $p \sim 2.4-2.6$) also shows notable distinctions in structure. It has a substantially stretched Cu-planar O_P distance d_{O_P} of 2.00 Å (normally around 1.92 Å), which may partially be due to the large Ba²⁺ ionic radius, and concomitantly occurring, yet very significantly compressed Cu-apical O_A distance d_{O_A} of 1.86 Å (compare 2.4 Å in La₂CuO₄). These values are substantially different from the isovalent (and isostructural) superconductor Sr₂CuO_{3+ δ} with up to $T_c = 95$ K [14–17], where d_{O_P} and d_{O_A} are nearly identical at 1.95 Å. Additionally, in Ba₂CuO_{3+ δ} reduction of oxygen concentration leads to a high fraction of vacancies in the O_P sites, not the O_A sites [4,5,8], whereas the superconducting system Sr₂CuO_{3+ δ} has vacancies in both O_P and O_A sites [17].

Several of these observations have been confirmed by theoretical calculations [18]. Large oxygen O_P vacancy

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concentrations break up the CuO₂ square lattices, leading to a one-dimensional (1D) Cu-O_P chain lattice [4,5,19]. Stoichiometric Ba₂CuO₃, i.e., the $\delta = 0$ phase, would have a pure 1D Cu-O_P chain lattice, but that stoichiometry has not been well investigated yet. The chain lattice in the sister compound Sr₂CuO₃ has been intensively investigated as an ideal 1D Heisenberg $S = \frac{1}{2}$ antiferromagnetic (AFM) system for over 20 years [20–24].

In this paper, we will investigate the $\delta = 0$ and $\delta = \frac{1}{4}$ phases through ab initio calculations. The latter one approximates the superconducting phase around $\delta \approx 0.2$. For the $\delta =$ 0 phase of Cu^{2+} , compared with those of Sr_2CuO_3 , the halffilled $d_{X^2-Y^2}$ orbital (notation to be established later) leads to half the interchain hopping strength but nearly identical intrachain hopping strength, suggesting this compound provides a more nearly ideal 1D Heisenberg $S = \frac{1}{2}$ AFM system, as will be discussed. Considering AFM order on the chain, the system is gapped by a 1D spin-Peierls instability at $q = 2k_F$. In contrast to the previous suggestion that the substantially compressed d_{O_A} leads to inverted crystal field splittings of the e_g orbitals [8], our calculations indicate that this system is not greatly affected by the unusual Cu-O separations. The energy gap, however, is substantially changed by variation of the Cu-O_P separation and by O_A phonon modes, likely related to the spin-Peierls instability.

The structures and computational methods we have used are described in Sec. II. In Sec. III, a description of the nonmagnetic $\delta = 0$ electronic structure is provided, followed by that of the magnetic (AFM) electronic structure and magnetic characteristics. Effects on the band structure of high-symmetry O displacements are presented in Sec. III D. Based on the inferred vacancies at the planar oxygen sites in $Ba_2CuO_{3+\delta}$, several lattices [12,18,25,26] with ordered planar oxygens have been suggested to approximate the superconducting phase. Among them, we have investigated in Sec. IV structures with brickwall [25] and two-type ladder plus chain lattices [18,26], which have Cu ions in up to three types of coordination. At $\delta = \frac{1}{4}$ doping, the bilayer structure of a ladder layer plus a chain layer is found to be the ground state among the three structures. This structure contains all three Cu coordinations, displaying both (roughly) half-filled and quarter-filled Cu orbitals. Copper oxidation states, e_g orbital characteristics, and characteristics of Cu coordination (four, five, and six) are compared to identify distinctive aspects of the favored bilayer structure.

In Sec. V the fermiology of the energetically favored bilayer structure is presented in more detail, to inform later experimental data. Section VI presents a discussion of our results, including some relationships to the long-standing studies of spin-half chains. A summary is provided in Sec. VII. Some additional results are presented in the Appendix.

II. METHODS AND STRUCTURES

A. Methods of calculation

We initially carried out first-principles calculations using the generalized gradient approximation (GGA) [27] for exchange and correlation, which is implemented in the allelectron full-potential code WIEN2K [28]. The results for the AFM phase at $\delta = 0$ were confirmed by another all-electron full-potential code FPLO-18 [29] (see the Supplemental Material (SM) [30]).

In WIEN2K, the basis size was determined by $R_{mt}K_{max} =$ 7 with augmented-plane-wave sphere radii R_{mt} (in a.u.): Cu, 1.71; O, 1.47; and 2.50 for Ba. For FPLO the default orbitals were used. For both codes, the Brillouin zone was sampled by a dense mesh containing up to 8000 k points to check the energetics and fine band structures (specifically, small energy gaps) carefully.

Structural optimization for the doped cases was carried out with the VASP code [31], using an energy cutoff of 600 eV. This code is efficient for the supercells, which contain up to 50 atoms. The lattice parameters were optimized, while the internal structural parameters were relaxed until all forces were less than 0.01 eV/Å (see SM for further information [30]).

Correlation effects of the electronic properties are always a consideration in cuprates. The most used procedure is to include Hubbard U and Hund's exchange J_H parameters in the DFT + U procedure [32]. We have done this for the insulating $\delta = 0$ compound. For the metallic and superconducting phase $\delta \sim 1/4$, calculations neglecting U give more realistic (sometimes excellent [33]) Fermi surfaces and energetics while excitations, which we do not address here, require dynamical calculations including U [34].

There is no single accepted method for calculating U, with various proposals depending on procedure, on the choice of "atomic orbital" (which might even be a Wannier orbital), and on the choice of treating the dynamical correlations. We discuss the commonly employed single-band expression for nearest neighbor exchange coupling of spins $J = 4t^2/U$ to provide a guide to the magnitude of magnetic coupling, and use U = 7 eV in quantities we provide, unless otherwise stated. For a range of values around this choice of U (which has been calculated [35–37] and applied with realistic results several times before) physical properties typically vary modestly with the choice of U.

B. Structures for $\delta = 0$ and $\delta = \frac{1}{4}$

The available structural information is ambiguous, and it seems that stoichiometric and ordered Ba_2CuO_3 (Cu²⁺) has not been achieved. Powder x-ray diffraction data on the $T_c = 70$ K multiphase sample was refined within the I4/mmmspace group (as for La₂CuO₄), obtaining [8] a = 4.00 Å, c = 12.94 Å, with 40% vacancies randomly placed on the *planar* O sites, obtaining the Ba₂CuO_{3,2} composition. The in-plane lattice constant is unexpectedly large compared to most layered cuprates, and an OA-Cu separation of 1.86 Å was emphasized as being exceptionally small, viz., 2.4 Å in La₂CuO₄. A cluster expansion study [26] of compositions around $\delta \sim 0.2$ found that ordered supercells containing Cu-O_P chains and ladders are favored up to 900 K. Our choices are guided by the chain structure of isovalent Sr_2CuO_3 for $\delta = 0$, and by the simplest choices of added in-plane oxygen atoms (described below) for $\delta = \frac{1}{4}$. These choices allow the assessment of differences due to coordination (there are four-, five-, and sixfold Cu ions), which seems to be a more useful viewpoint than crystal field splitting.



FIG. 1. (a) Idealized crystal structure of Ba₂CuO₃, based on the 1D Cu-O_P chain. The apical oxygens O_A and Ba ions sit at nearly identical heights. The Cu-O_P chains are well separated by the insulating BaO_A layer, by 6.57 Å along the \hat{c} direction. (b) Spin density plot (GGA) of the Ba₂CuO₃ AFM state, showing the strong $pd\sigma$ hybridization. The isovalue is 0.015 $e/Å^3$. Each spin character is described by the different colors.

 $\delta = 0$: Ba₂CuO₃. Our first calculations are based on the lattice parameters of a = b = 4.003 Å and c = 12.942 Å [8], recently refined in a superconducting sample at room temperature as noted above. These values are close to the early reported values [4,5,19] also on multiphase samples. For Ba₂CuO₃ with 50% ordered vacancies on the planar oxygen site (space group *Immm*, No. 71), shown in Fig. 1(a), the Cu and O_P ions sit at 2a (0,0,0) and 2d (0, $\frac{1}{2}$,0) sites, respectively. The Ba and O_A ions lie on $4i(0,0,\xi)$ sites with internal parameters $\xi_{Ba} = 0.1437$ and $\xi_{O_A} = 0.1438$, at effectively identical heights. As discussed below, calculations for Ba2CuO3 show a substantial tetragonal $(a \neq b)$ distortion, reflecting the chain structure. This difference is consistent with the experimental observations in $Sr_2CuO_{3+\delta}$ that show a rapidly reduced difference in the planar lattice parameters, as oxygen concentration in Sr₂CuO_{3+ δ} is increased from $\delta = 0$ [17].

We began our investigations from the $\delta = 0 \text{ Cu}^{2+}$ phase that serves as the closest stoichiometric parent compound of the Ba₂CuO_{3+ δ}. We use the chain structure originally proposed by Armstrong *et al.* [3]. In contrast to the abovementioned *I4/mmm* optimization of the 70 K sample, other data on Ba₂CuO_{3+ δ} has indicated a difference of in-plane lattice parameters, less than 1% (0.04 Å) in a high temperature, possibly strongly site disordered, sample [5,6,19] but a large 0.3 Å difference in a low-temperature sample [4]. This tetragonal distortion is consistent with ordering of O vacancies, likely in a 1D Cu-O_P chain (sub)structure. Due to the differing reports, for the $\delta = 0$ phase, we fully optimized all structural parameters in the orthorhombic space group *Immm*.

 $\delta = \frac{1}{4}$: Ba₂CuO_{3.25}. To describe models of ordered planar oxygens in the superconducting doped regime, we investigated three structures of doubled or quadrupled supercells (as necessary) having two or three Cu sites, pictured in

Fig. 2 [18,25,26]. All structures are derived from the Ba₂CuO₃ layers of Cu-O_P chains in which each Cu has an apical O above and below. The structures differ in where the additional planar O, necessarily bridging chains, per four Cu sites is distributed. The additional ions produce fivefold coordinated Cu sites (denoted Cu5) along the chains, octahedral sixfold (Cu6) sites in the bilayer structure, and new fourfold sites (Cu4*) on the rungs in addition to the Cu-O_P chain site (Cu4), which is the only site for $\delta = 0$. We describe the structures specifically because the coordination has strong effects on the Cu e_g orbitals.

Bilayer structure. Figure 2(a) illustrates the oxygen ladder ordering, which consists of chains bridged by O-Cu-O units. In this bilayer case the $z = \frac{c}{2}$ and z = 0 layers consist of the Cu-O ladders and Ba₂CuO₃-type chain lattices, respectively. In this structure two bridging O ions are inserted into alternating layers. The $z = \frac{c}{2}$ layers can be viewed as connected ladders. There are one sixfold coordinated Cu ion (Cu6) and three fourfold Cu ions, one along each chain (Cu4) and one along the rungs (Cu4^{*}).

Monolayer structure. In this structure the two layers are identical (hence "*monolayer*"), but displaced by a body-centering operation. The additional O ions (with respect to Ba_2CuO_3) are shown in Fig. 2(b) to lead to a chain and a ladder. There are two fivefold coordinated Cu5 sites, and two distinct fourfold sites, Cu4, along the chains and Cu4* forming the rung of the ladder.

Brickwall structure. This structure [25] consists of one layer of Cu-O chains (Cu4 sites), two per primitive supercell. The other layer has a brickwall ordering of oxygen vacancies, as illustrated in Fig. 2(c). Planar oxygen vacancies occur every two squares along one in-plane direction, being staggered by (a, a, 0) in the other direction to give the brickwall motif. Each copper site in this layer is bonded to two apical O ions and three planar O ions, thus being fivefold coordinated Cu5 sites.

III. RESULTS FOR PARENT $\delta = 0$ PHASE

A. Structural items

Structural differences between nonmagnetic (NM) and AFM states were on the order of 10^{-2} Å. The optimized NM lattice parameters are a = 4.144, b = 3.808, c = 13.142 (in units of Å), with the internal parameters of $\xi_{Ba} = 0.1459$, $\xi_{O_A} = 0.1481$. The corresponding Cu-O_P and Cu-O_A distances are 2.07 and 1.95 Å, respectively, a difference of $\pm 6\%$. The lattice parameters are consistent with the early experimental low-temperature values [4]. As expected, the Cu-O_P chains lie along the longer \hat{a} direction. Compared with the experimental value of the superconducting sample [8], the Cu-O distances are larger by 0.07-0.08 Å, but the interchain distance is smaller by 0.2 Å in the layer; very significant differences. For our optimized structures with the same cell volumes [38], the $a \neq b$ case has a lower energy by 30 meV/f.u. than for a = b. Thus for the $\delta = 0$ we will focus on results for the optimized $a \neq b$ structure, unless specified otherwise.

In both a = b and $a \neq b$ cases, the AFM state with alternating spin alignment along the Cu-O_P chain has a lower energy than in the NM state, by 4 meV/f.u. for the $a \neq b$ case.



FIG. 2. Crystal structures, investigated here, for the $\delta = \frac{1}{4}$ phase. For simplicity, only Cu (larger spheres) and O (smaller, red spheres) ions are shown. Both the bilayer (a) and brickwall (c) structures have 1D Cu-O chains extending horizontally in the z = 0 layer, whereas the monolayer (b) phase has the same structure in both the z = 0 and the $z = \frac{c}{2}$ layers. Each structure is derived from the Ba₂CuO₃ chain structure, by (i) inserting O ions linking the chains, and (ii) moving the planar O on alternating chains to complete the ladder rung, or the long edge of the brick. We used supercells of $2 \times 1 \times 1$ for (a), $1 \times 4 \times 1$ for (b), and $2 \times 2 \times 1$ for (c). Here, the \hat{a} direction is always chosen along the Cu4-O_P chain line.

This difference is remarkably small, but comparable with the energy gain of about 7 meV/f.u. from a simple Stoner picture $I_{st}M^2/4$ with Stoner $I_{st} = 0.9$ eV [32] and the Cu moment of $0.18\mu_B$ (see below). This indifference of the AFM energy gain to the a/b ratio is another property that is insensitive to the tetragonal distortion in the planar lattice parameters. Our attempts to obtain a ferromagnetic state always reverted to NM states.

toward the apical oxygen O_A . Consistent with the $Cu^{2+} d^9$ configuration, the $d_{X^2-Y^2}$ antibonding state with a band width of 2.3 eV crosses the Fermi energy E_F exactly at half filling. The less dispersive d_{Z^2} orbital (perpendicular to the CuO₄ unit) is occupied at -3 eV. The half-filled $d_{X^2-Y^2}$ band leads to a strongly 1D Fermi surface (FS), displayed in Fig. 3(b), suggesting strong 1D instabilities at $q = 2k_F$ (see below).

B. Nonmagnetic electronic structure

Now we address the NM state to establish the underlying (spin-symmetric) electronic structure. The bands and corresponding atom- and orbital-projected densities of states (PDOSs) are provided in Figs. 3(a) and 4, respectively. These are similar to the previously reported ones in the a = bcase [13]. To analyze this, we focus on the CuO₄ unit and choose the X axis along the Cu-O_P chain and the Y axis



FIG. 3. (a) GGA nonmagnetic band structure for the optimized structure, with fatband representation of the Cu $d_{X^2-Y^2}$ and d_{Z^2} characters in the range of -6 eV to 3 eV, containing the Cu 3*d* and O 2*p* orbitals Ba₂CuO₃. The *X* and *Y* points are chosen as the zone boundaries parallel and perpendicular to the Cu-O_P chain, respectively. (b) The corresponding Fermi surfaces, showing very strong 1D character arising from the Cu-O_P chain band.



FIG. 4. For NM Ba₂CuO₃, (top) total and atom-projected densities of states, and (bottom) orbital-projected densities of states for each atom, within GGA. The total Fermi level DOS $N(E_F) =$ 0.72 eV⁻¹ per f.u. for both spins. The total DOS, yellowish-shaded in the top panel, varies little around E_F , reflecting the midband 1D character. Due to the chain structure, the mostly filled O p_{π} orbitals show differences between the two sites.

Figure 3 suggests that a simple single-band model is a reasonable starting point for Ba₂CuO₃. Fitting the $d_{X^2-Y^2}$ band leads to the following nearest neighbor (NN) and next NN hopping parameters:

- (1) a site energy of -0.146 eV;
- (2) an intrachain hopping $t_a = 0.532 \text{ eV}$;
- (3) a second NN intrachain hopping $t'_a = 0.083 \text{ eV}$;
- (4) an interchain hopping $t_b = 0.044 \text{ eV}$;
- (5) a second neighbor hopping $t_{ab} = 0.015$ eV;
- (6) an interlayer hopping $t_c = -0.010$ eV.

Keep in mind that these values correspond in principle to a Cu-centered Wannier function with $d_{X^2-Y^2}$ symmetry and character. Compared with our calculated hopping parameters of Sr₂CuO₃ with the experimental structure parameters of Ami *et al.* [20], t_b is reduced by half; changes in the other parameters are unimportant. This comparison indicates that Ba₂CuO₃ is an even more nearly ideal 1D Heisenberg AFM spin- $\frac{1}{2}$ chain system than Sr₂CuO₃ (see below).

A first interest in magnetic insulators such as this is the exchange couplings between magnetic ions. These hopping parameters indicate several exchange couplings $J_a \equiv J$, J'_z , J_b , J_{ab} , and J_c . The dominant intrachain hopping t_a is conventionally used to estimate a corresponding superexchange strength $J_a = 4t_a^2/U$. For the value of U = 7 eV that we have used in the following calculations, one obtains $J_a \approx 0.162$ eV (i.e., about 1875 K). With the value U = 8 eV from a spectroscopic (high-energy) experiment [10], the estimate reduces accordingly, to 1640 K. There have been suggestions that Uas small as 5 eV might be realistic (at least in square-lattice cuprates), leading to a considerably larger value of J_a .

It is useful to keep in mind the origins of the uncertainty in choosing, or fitting to experiment, a single value of NN J. The bigger picture is that the superconducting composition $\delta \sim \frac{1}{4}$ is the prime interest, and our calculations below indicate the importance of the d_{z^2} orbital, so a single-band picture no longer holds. Even when the additional parameters (site energies and crystal field splitting, charge transfer energy, altered hopping amplitudes) become available, the various exchange couplings become challenging to obtain, with fourth-order perturbation theory expressions not being accurate [39]. There are the usual complications, that a high-energy value of U (photoelectron spectroscopy) is different [40] from lowenergy values (transport, spin waves), and also that U values may be different for the two e_g orbitals [41]. Other differences in principle also arise, such as that DFT + U codes nearly always deal with (U corrections to) atomic orbitals, whereas model Hamiltonians conventionally neglect overlap integrals, thus assuming Wannier functions as the basis. Our parameter values (above) provide the starting point for more elaborate model studies of $Ba_2CuO_{3+\delta}$.

Focusing on the exchange parameters, a conventional practice when there are stable localized moments is to compare the energy differences for different alignments of the moments calculated using the appropriate functional (usually DFT + Urather than DFT alone). AFM versus FM alignment gives an estimate of the NN coupling (here, J_a). Extending to larger supercells, other lower symmetry alignments provide more energies, and a number of exchange constants can be fit to them. We have found that, for the value U = 7 eV for which we report results, no FM can be obtained; the moment iterates



FIG. 5. AFM band structure of optimized Ba₂CuO₃ within GGA, highlighting the spin-resolved characters by fatband plots of the Cu $d_{X^2-Y^2}$ and d_{Z^2} characters with a positive moment. The AFM-induced gap indicates insulating character. The *S* point is the zone boundary in the (110) direction, as described in Fig. 9.

to zero. Since the AFM state (and also no FM state) was obtained even for U = 0 (see below), the lack of FM solution confirms that the AFM moments we do obtain are not of primarily local character. Also, their magnitudes are smaller than expected of spin- $\frac{1}{2}$ moments, even accounting for hybridization with oxygen. A modest FM moment would result in exchange split bands, which from Fig. 3(a) would leave a metallic, not insulating, electronic structure, uncharacteristic of single-band local moments. A reliable means of obtaining the exchange parameters would be to perform self-consistent linear response calculations of the spin-wave spectrum [42], which would be very useful if stoichiometric Ba₂CuO₃ can be synthesized and characterized.

As mentioned above, effects of the tetragonal distortion of the *a-b* lattice parameters on the electronic structures and magnetic characters are not substantial. However, a few distinctions can be observed. The distortion decreases t_a by 10% and increases t_b by a factor of $\frac{3}{2}$, reflecting the increased intrachain and reduced interchain Cu-O_P separations. In both NM and AFM states, a clearly distinguishable change is to increase a gap between the E_F -crossing $d_{X^2-Y^2}$ band and the bottom of the unfilled bands (the *M* point) at 1 eV.

C. Antiferromagnetic state: Spin-Peierls instability

The AFM state, which is the expected and calculated ground state of Ba₂CuO₃ [18], was investigated using a supercell doubled along the Cu-O_P chain direction. The band structure in GGA is displayed in Fig. 5, showing an AFM-induced gap [43] of 0.2 eV. The half-filled $d_{X^2-Y^2}$ band lies in the range of -1.5 to 1 eV, whereas the fully filled d_{Z^2} orbital lies around -2.5 to -3 eV. The corresponding orbital PDOSs are given in Fig. 6, indicating strong $pd\sigma$ hybridization, consistent with the spin density plot of Fig. 1(b). The Cu spin moment, ideally a spin- $\frac{1}{2}$ value of $1\mu_B$, is strongly reduced to $0.18\mu_B$ by the $pd\sigma$ hybridization. The O_P ions are thereby strongly polarized, but have zero net moment due to AFM symmetry.

A noteworthy aspect in the AFM state is a gap of 0.2 eV opening at the GGA level without the assistance of a Hubbard



FIG. 6. GGA orbital PDOSs of Cu, planar, and apical oxygens in AFM Ba_2CuO_3 . The Cu ion with a positive moment is chosen. Due to AFM symmetry, the only spin-polarized holes in the oxygens are on O_A .

U repulsion, which is required in square-lattice cuprates. Gap opening without U eliminates the categorization as a Mott insulator, leaving the characterization of Ba₂CuO₃ as in the Slater insulator regime, as far as gap opening and the origin of magnetism are concerned. The zone folding due to doubling of the cell along the \hat{a}^* direction ($a^* = 2a$) leads to overlapped bands along the X'-S line, i.e., on the $k_x = \pi / a^*$ plane. Then nesting of the 1D FS, evident in Fig. 3(b) that it is near perfect, leads to a 1D spin-Peierls instability, analogous to the lattice-Peierls instability-symmetry breaking along the chain is the underlying mechanism in both cases. As shown in the band structure of Fig. 5, along the X'-S line the $d_{X^2-Y^2}$ bands bordering the gap have opposite spin character. That is not the complete story. As in other cuprates, the Hubbard Usurely has a strong effect, increasing the gap and moment in the insulating phase and impacting spectral density spectra in the metallic regime.

Inclusion of the Hubbard U to the Cu ions simply increases the energy gap and the magnetic moment, since the gap is already open in the GGA level. At U = 7 eV, the moment of the Cu ion reaches $0.64\mu_B$ and the gap is about 1.5 eV.

D. Effects of oxygen mode displacements

We have performed frozen phonon calculations for a few $\vec{Q} = 0$ O displacements, two involving O_A motion and one involving O_P motion. One is the beating A_g mode of O_A ions against the Cu-O_P chain with amplitudes up to ± 0.06 Å. The change in energy versus displacement (not shown here) is fit well with a simple harmonic form, with frequency of 603 cm⁻¹, which is within 7% of the observed Raman frequency [7] of 563 cm⁻¹ in a multiphase sample of Ba₂CuO_{3+ δ} with δ in the 0.09–0.17 range.

We have studied changes with an increase of the displacement in the IR-active B_{1u} mode of the O_A-Cu-O_A unit. The energy gap linearly reduces with O displacement, reaching $\Delta E_g = -0.1$ eV at the amplitude of ± 0.06 Å that is a typical rms displacement of an oxygen ion (see SM [30]). The frequency of the B_{1u} mode is 560 cm⁻¹.

The O_P breathing mode along the chain direction was studied for amplitudes u_0 up to 0.07 Å. This broken symmetry results on two Cu sites alternating along the chain, but the allowed spin disproportion is negligible, i.e., spin-lattice coupling is small. This oxygen optic mode has an anomalously low energy frequency of roughly 43 cm⁻¹, probably a remnant of the Peierls instability. The electron-lattice coupling shows up in the modulation of an energy gap, with $dE_g/du_0 \approx$ 3.0 eV/Å. The gap vanishes at 0.07 Å; since this is a typical oxygen rms vibrational amplitude, electron-phonon coupling in Ba₂CuO₃ may be worth further study.

IV. RESULTS FOR THE SUPERCONDUCTING PHASE

Before considering structures of ordered planar oxygens in the superconducting phase around $\delta \approx 0.2$, we carried out calculations of the virtual crystal approximation (VCA) up to $\delta = 0.25$ by increasing the number of valence electrons from the $\delta = 0$ phase. As might be expected from the AFM DOS given in Fig. 6 and the FS $2k_F = \pi/a$, once hole-doped the system moves away from the spin-Peierls instability and the gap and the small moment vanish. In this region, the single-band feature at $\delta = 0$ remains unchanged from the NM character (not shown here). Above $\delta \approx 0.25$, a small AFM moment reappears as the increased hole density begins to favor polarization.

A. Supercells

We now turn to the supercell results at $\delta = \frac{1}{4}$. The bilayer structure is favored, with the monolayer cell being 70 meV higher and the brickwall structure being 250 meV higher than the bilayer [44]. These are of course groundstate (zero temperature) values. For varying high-pressure and high-temperature synthesis conditions, samples might contain regions with the metastable monolayer structure interspersed with a majority amount of bilayer domains. (Our band structure of the monolayer structure, shown in SM [30], is similar to that published previously [26].)

Regularities across these structures based on coordination can be identified. The PDOS spectra of the various Cu sites are provided in Fig. 7 and the Appendix. In all three structures:

(1) The *fourfold coordinated* Cu4 chain sites display filled and strongly bound Z^2 orbitals while the $X^2 - Y^2$ bands are roughly half-filled. This filling is characteristic of a Cu²⁺ ion with large crystal field splitting. Isolated chains in each structure are likely to have strong AFM correlations, which (in the absence of magnetic ordering, which is not yet reported) will reduce but not eliminate conduction along the chain (see Sec. V for further description).

(2) The *fivefold* Cu5 sites in the two disfavored structures (see the Appendix) show holes distributed between both e_g orbitals, reflecting the vast reduction in the e_g crystal field splitting compared to Cu4 sites. In the brickwall structure, the hole count is dominated by $x^2 - y^2$ weight (z^2 weight is minor); in the monolayer structure, surprisingly, z^2 hole weight dominates. In each case the amount of Cu5 holes seems roughly characteristic of a formal valence of 2.5+.



FIG. 7. The Cu e_g orbitals' projected DOSs for the NM $\delta = \frac{1}{4}$ bilayer structure pictured in Fig. 2(a), within GGA. For the differing types of local coordination, see the text.

(3) For the *fourfold rung* Cu4^{*} site, the hole(s) lies entirely in the roughly half-filled $X^2 - Y^2$ band, thus with one hole of each spin, again close to a nominally Cu²⁺ ion, as for the chain Cu4 ion. In the monolayer structure (see the Appendix), the Cu4^{*} DOS is gapped and clearly 2+. Differing somewhat from the brickwall structure, both Cu4 and Cu4^{*} ions in the bilayer case have strongly bound Z^2 states, with metallic but different from half-filled $X^2 - Y^2$ bands.

(4) With two (chain layer) $Cu4^{2+}$ ions in the bilayer primitive cell, the average formal valence of the Cu6 and Cu4* ions must be $Cu^{2.5+}$. The PDOSs in Fig. 7 reveal that the *sixfold* Cu6 ion has *more* z^2 *holes than* $x^2 - y^2$ *holes*, making it very highly oxidized (viz., approaching $Cu^{2.75+}$) and, given the overall highly hole-doped character, a likely candidate to be heavily involved in superconducting pairing and in hosting the superconducting carriers. Supporting this viewpoint is the expectation that strong spin correlations along the chain will reduce the participation of the Cu4 ion. An interesting parallel is that z^2 holes have also been implicated [45,46] in the long sought and recently discovered superconductivity in infinitelayer hole-doped NdNiO₂ [47].

B. The bilayer oxygen ordering

We now focus on this energetically favored bilayer structure, whose electronic features have not been investigated previously. In this phase, our calculations in WIEN2K were performed with the experimental lattice and internal parameters given in Ref. [8].

In the bilayer structure of Fig. 2(a), the supercell, containing four formula units, has two Cu4 (chain) sites in the z = 0 plane, and one planar Cu4^{*} (rung) CuO₄ unit and one sixfold coordinated Cu6 in the $z = \frac{c}{2}$ plane. In this structure, our attempts to obtain an AFM state (after all, in one layer two well-isolated chains remain intact) in GGA always reverted to a NM state. This result might be thought to be consistent with the previous calculation [18], and our VCA results, if charge transfer between layers occurs. Since the material is (super)conducting, we did not pursue possible magnetic and insulating states using correlated electronic structure methods.

Figure 7 shows the e_g orbitals' PDOSs for all three Cu sites. Although the electronic structure is metallic, a formal valence viewpoint might still be instructive. Since the Cu4 and Cu4* Z^2 states are both localized near -3 eV, their nominal valences (from $X^2 - Y^2$ hole count) should be nearly the same. On the contrary, the Cu6 sites contain hole(s) from both orbitals, with (surprisingly) somewhat more z^2 weight than $x^2 - y^2$ weight. Figure 7 indicates that their crystal fields are drastically different: that of Cu6 is close to an octahedral e_g degeneracy.

C. Charge, crystal fields, bands

Charge density. Bader charges can sometimes provide insight. They do depend on environment and thus are not true atomic properties, and often do not correspond closely to formal valence (atomic sphere charges also can vary greatly from formal valences). Differences in Bader charge for inequivalent ions give the most useful information. Bader charges of the Cu ions are +1.31 (Cu6), +0.89 (Cu4^{*}), and +0.86 (Cu4). The similarity of the latter two supports equal formal valence, and the difference of 0.42 to 0.45 relative to Cu6 indicates a substantially different valence for Cu, again in agreement with the analysis above based on PDOSs. The charges of the apical oxygens are -1.12 to -1.28, while those of the planar oxygens are -1.08 to -1.11. These differences of 0.03 to 0.2 likely reflect primarily differences in environment (bonding).

Crystal fields. As expected from a simple crystal field concept, the site energy of the d_{z^2} orbital (equal to that of $d_{x^2-y^2}$ for an ideal octahedron e_g partner) changes with variation of the Cu-O_A distance d_{O_A} . We find that the d_{z^2} orbital remains partially unfilled in the range of d_{O_A} distances from 1.86 to 2.10 Å. The center of the d_{z^2} band shifts by at most 0.1 eV in this range. These results indicate that the effects of the shorter Cu-O_A distance on the electronic structure are not substantial, consistent with the results of Ba₂CuO₃. In addition, as seen above in Ba₂CuO₃, the corresponding variation of the total DOS around E_F is negligible (not shown here). This indifference to d_{O_A} suggests that this bond length variation with doping has little effect on T_c . At E_F the total DOS is low and nearly identical to that of the NM $\delta = 0$ phase, $N(E_F) = 0.73 \text{ eV}^{-1}$ per f.u. for both spins.

Band structures. The corresponding band structure enlarged in the -2 to 2 eV region is shown in Fig. 8, with the highlighted fatband plots of Cu6 $d_{x^2-y^2}$ and d_{z^2} orbitals, and Cu4* and Cu4 $d_{X^2-Y^2}$ orbitals. Note first the lack of dispersion along Γ -Z—the layers remain uncoupled, except for possible charge transfer. The partially filled $d_{X^2-Y^2}$ orbitals of Cu4* and Cu4 ions show very similar 1D dispersion, with widths of ~2.4 eV, leading to $t_a \approx 0.6$ eV. The Cu4* orbital shows d-dhybridization with Cu6 e_g orbitals near E_F along the S- Γ line, due to accidental degeneracy and some small coupling. On the other hand, the Cu4 orbital is negligibly hybridized with the other d orbitals of Cu6 and Cu4*, reflecting the 1D character of the chain.

Both (antibonding) Cu6 $d_{x^2-y^2}$ and d_{z^2} orbitals of width 2.3 eV cross E_F and are energetically nearly degenerate. These Cu6 bands show two-dimensional (2D) character with



FIG. 8. Band structure, enlarged in the range of -2 eV to 2 eV, of the NM $\delta = \frac{1}{4}$ bilayer phase, within GGA. The characters of the e_g orbitals of (a) Cu6, (b) Cu4*, and (c) Cu4 are highlighted by magenta $(d_{x^2-y^2} \text{ or } d_{X^2-Y^2})$ and green (d_{z^2}) colors.

 $t_a \approx t_b \approx 0.56$ eV. As a result, within a formal charge picture, the Cu4 and Cu4* ions are Cu²⁺ with half-filled $d_{X^2-Y^2}^1$ bands. As noted, for formal charge balance, the Cu6 ion must be overdoped Cu^{2.5+}. Formal charges can be questionable for metallic systems, of course, but the consistency of identification is striking. This difference in Cu sites indicates that the ordered planar oxygen leads to a mixture of single-band and two-band features that could significantly complicate determination of the superconducting pairing symmetry and mechanism.

V. FERMIOLOGY OF THE BILAYER STRUCTURE

Figure 9 shows the Fermi surfaces of the NM bilayer $\delta = \frac{1}{4}$ phase. All FSs show strongly reduced dimensionality. They contain a large rounded-square barrel surface centered at Γ , with negligible dispersion in the third direction, analogous to several doped cuprate compounds. In addition there are quasi-1D sheets near the X' - S zone edges, one extending along the entire edge length, the other cut off midway by small in-plane



FIG. 9. Fermi surfaces within GGA of the NM $\delta = \frac{1}{4}$ phase in the bilayer structure, with several flat parallel sheets tending to support charge, spin, or lattice instabilities.

dispersion. These surfaces are indicative of strong 2D and 1D characters in the respective bands. The character of the barrel FS is a mixture of the Cu6 e_g orbitals with the Cu4* $d_{X^2-Y^2}$ orbital. The 1D FSs have mostly chain Cu4 and Cu4* $d_{X^2-Y^2}$ character, with minor admixture of Cu6 $d_{x^2-y^2}$.

The rms Fermi velocities over all sheets are

$$v_{F,a} = 24.7, \quad v_{F,b} = 4.5, \quad v_{F,c} = 0.05$$

in units of 10^7 cm/s, a factor of 5 in planar anisotropy and likely at most hopping conductivity in the perpendicular direction. These large velocities account for the rather small Fermi level DOS in spite of large FSs. This strong anisotropy is also reflected in the Drude plasma energies (in eV) of

$$\Omega_a = 3.9, \quad \Omega_b = 1.66, \quad \Omega_c = 0.17,$$

but smaller in magnitude. In contrast to the two 1D FSs containing hole carriers with $v_{F,a}^{1D} \approx 34.5 \times 10^7$ cm/s, the Γ -centered barrel FS is less anisotropic in-plane:

$$v_{F,a}^b = 7.2, \quad v_{F,b}^b = 11.1, \quad v_{F,c}^b = 0.08$$

in units of 10^7 cm/s. This corner-rounded square FS with a side of $0.35\frac{\pi}{a}$ possesses roughly 0.50 electrons, while hole carriers near the zone boundary are contained within the 1D FSs.

One can note that this superconductor has a low $N(E_F)$ and low carrier density, a characteristic shared with other superconducting cuprates. This is, however, different from the superconducting Fe pnictides with their higher $N(E_F)$ together with low carrier density [48]. However, several superconducting cuprates show (within GGA) low $N(E_F)$ along with higher carrier density [33].

VI. DISCUSSION

Bonner and Fisher demonstrated that quantum fluctuations preclude magnetic ordering of the (isolated) spin- $\frac{1}{2}$ Heisenberg chain at finite temperature [49], although strong spin-spin correlations lead to a vanishing susceptibility as zero temperature is approached. Nevertheless, in the presence of even very small interchain coupling, magnetic ordering is observed to occur in several spin-chain materials. We have quantified the interchain coupling in Sec. III.

Schulz treated these intrachain correlations exactly and interchain coupling in mean field, and provided a formula for the staggered magnetization $m_0 = \langle S_i^z \rangle$ at the *i*th site and for the Néel temperature T_N of the 1D Heisenberg AFM spin- $\frac{1}{2}$ chain [50]. For Ba₂CuO₃, the staggered magnetization is

$$m_0 \approx g\mu_B \times 1.02 \frac{t_b}{t_a} \approx 0.17 \mu_B,\tag{1}$$

with g being the Landé g factor. This result indicates the degree to which $1\mu_B$ is strongly reduced by quantum fluctuations. Our calculations indicate that Cu-O $pd\sigma$ hybridization reduces the *static* ordered moment from $0.2\mu_B$ to $0.6\mu_B$ depending on the size of Hubbard U, which will be further reduced by fluctuations. The magnetic susceptibility should contain signatures of the strength of spin correlations as well as magnetic ordering.

Schulz also provided an implicit equation for ordering temperature T_N in terms of simple interchain coupling J_{\perp} and two

constants A = 0.32, $\Lambda = 5.8$ from numerical calculations:

$$|J_{\perp}| = \frac{T_N}{4A\sqrt{\ln(\Lambda J/T_N)}}.$$
(2)

With our estimates from Sec. III J = 1640 K, $J_{\perp} \sim 10$ K, this expression gives T_N in the neighborhood of 34 K. His expression with a single interchain coupling does not fit closely with the Ba₂CuO₃ structure with several (small) interchain couplings that we obtain, so we have used the largest. Within the uncertainty of the exchange couplings and the logarithmic dependence of the expression, this is similar to but somewhat larger than the case of Sr₂CuO₃ [24], which means that Ba₂CuO₃ is also near the regime of the 1D Luttinger-liquid quantum-critical phase.

In the superconducting sample [8], the linear specific heat coefficient $\gamma \sim 14 \text{ mJ/mol } \text{K}^2$ was experimentally measured. Comparison of this value to our calculated band value of $\gamma_0 = 1.72 \text{ mJ/mol } \text{K}^2$ leads to a factor of 8 due to dynamic correlation effects, implying a strong correlation strength in the doped material.

Our calculations indicate that this system is little affected by the unusual Cu-O separations. This behavior can be contrasted with the infinite layer Ni²⁺ system of Ba₂NiO₂(AgSe)₂, in which the strained Ni-O separation leads to a calculated "off-diagonal singlet" state [45], consistent with previous findings by some of the current authors that Cu ions are different from Ni ions even if both have the same ionic configuration [51,52].

The bilayer FSs suggest instabilities of the ordered structure that may help in understanding the paucity of singlephase, well-ordered samples. Flat (or parallel—nesting) regions of FSs provide a strong tendency toward electronic instability that manifests as charge, spin, or lattice instabilities, each of which break the symmetry of the underlying phase. The barrel FS shows nesting at $2k_F^b \sim 0.70\frac{\pi}{a}$ in both of the planar directions. The 1D FSs suggest instabilities at a small value of $2k_F^{\text{ID}}$ in the \hat{a} direction (i.e., a long-wavelength disturbance), and a large value $2k_F^{\text{ID}} = \frac{2\pi}{a} - 2k_F^{\text{ID}}$, which corresponds to a short-wavelength disturbance. These potential or real instabilities may help to account for the complex structures observed in samples.

VII. SUMMARY

Our first-principles studies of the electronic and magnetic structures of Ba₂CuO_{3+ δ}, $\delta = 0$ or $\frac{1}{4}$, provide important information about this unusual superconductor. Our studies indicate that interchain coupling in NM (hence metallic) Ba₂CuO₃ is quite small, thus not surprisingly, this state is subject to Peierls-type instability of spin, of lattice, and possibly also of charge ordering. Considering other cuprates, we have focused on the spin (AFM) instability. AFM ordering alleviates the Peierls instability; however, the magnetism is weak without the assistance of intra-atomic Cu 3*d* repulsion and correlation. Charge and lattice instabilities would compete with AFM ordering. If AFM ordering occurs, it will be weak due to the fact that interchain coupling is very small, and isolated chains do not order—the Bonner-Fisher result. Considering the very stable Cu²⁺ ion it is surprising that sto-

ichiometric, ordered Ba₂CuO₃ has not been obtained, even in samples synthesized with varying pressure and temperature.

Isolated 1D spin- $\frac{1}{2}$ chains do not order due to strong fluctuations; however, even small interchain coupling does lead to order. The large intrachain and small interchain couplings that we estimate, using the theory of Schulz, suggest an ordering temperature of the order of 34 K. This value puts Ba₂CuO₃ in the regime of the 1D Luttinger-liquid quantum critical phase, for which further experimental study will be useful.

The types of ordering of the planar oxygen dopants play a crucial role in determining the crystal and electronic structures of Ba₂CuO_{3+ δ}, and thereby the value of *T_c* and pairing mechanism. This observation is consistent with experimental data that show substantial change of *T_c* related to the modulation of apical oxygen positions and concentration in superconducting Sr₂CuO_{3+ δ} [16,17]. The doping level of $\delta = \frac{1}{4}$ has led to the picture of ordered approximants being modeled as additional O ions being inserted, seemingly necessarily bridging Cu-O_P chains, rather than removing a much larger fraction of O ions from Ba₂CuO₄.

Our study of three proposed ordered structures of $Ba_2CuO_{3+\delta}$ has indicated that a *bilayer* structure (one with two structurally different layers) is most stable, although a *monolayer* ordered structure is only modestly higher in energy. In the bilayer structure, all three Cu ions have e_g states at the Fermi energy and participate in conductivity, hence presumably also in superconductivity. There are, however, strong site differences: the two fourfold coordinated sites have the Z^2 orbital strongly bound and localized, leaving only more or less half-filled $X^2 - Y^2$ holes as occurs in most cuprates.

The octahedrally coordinated site, which occurs only in the bilayer structure, has a large fraction of holes in both e_g orbitals, with more in the z^2 orbital. This marks this site as very highly oxidized (viz., 2.75+), accounting for the perplexing high mean formal valence of Ba₂CuO_{3+ δ}. A quasidegeneracy of the e_g orbitals is restored by the octahedral coordination, and implies that Ba₂CuO_{3+ δ} is not only strongly overdoped but also a two-band superconductor.

In addition, this energetically preferred bilayer structure displays intense Fermi surface nesting of both 1D and 2D nature that will drive the system toward instabilities of a few types (wavelengths) of charge-, spin-, and lattice-symmetry breaking. The fluctuation associated with these instabilities may help to account for Ba₂CuO_{3+ δ} not having been prepared in a fully ordered, single-phase sample even for a variety of synthesis conditions.

The current work on the metallic phase has not dealt specifically with correlation effects due to strong repulsion in the 3*d* shell of the Cu ion. Crystal fields separating the two e_g orbitals on the fourfold and sixfold coordinated Cu sites have been identified and analyzed, and formal valence identifications have been discussed. Specifically, fourfold sites qualify as 2+, with the others better characterized as 2.5+ or even somewhat more highly oxidized. Our results provide a roadmap for further analysis and experimental exploration of the electronic structure of Ba₂CuO_{3+ δ}.

Note added. Recently, a dynamical mean-field work similar to this appeared in Ref. [54]. In their study they used U = 3 eV, but for the superconducting phase, while we treat only the insulating phase with DFT + U.



FIG. 10. The Cu e_g orbitals' PDOSs for the NM $\delta = \frac{1}{4}$ (left panel) monolayer structure pictured in Fig. 2(b) and (right panel) brickwall structure pictured in Fig. 2(c), within GGA.

ACKNOWLEDGMENTS

H.S.J. and K.W.L. were supported by National Research Foundation of Korea Grant No. NRF-2019R1A2C1009588. W.E.P. was supported by NSF Grant No. DMR 1607139.

APPENDIX

Since the PDOSs and associated crystal field splitting of the Cu e_g orbitals play a strong role in our identification of the Cu sites and orbital character that provide the superconducting carriers, we provide here, for the $\delta = \frac{1}{4}$ phase, the Cu e_g orbitals' PDOSs for the energetically unfavored structures in

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Fig. 10. Strong differences can be seen in the d_{z^2} spectra: for the Cu5 site, there is strong character at and above E_F , while that orbital is localized at -3 eV in the two fourfold sites, while great similarities are evident. It is not surprising that the "isolated" chain site Cu4 spectrum is relatively undisturbed by doping on other layers as in Ba₂CuO₃. There is, however, a strong difference between Cu4 and Cu4*: the former is half-filled, while the latter is gapped around E_F , seemingly approaching a 3+ formal valence. The brickwall Cu5 site (the right panel of Fig. 10) likewise approaches this same high formal valence. The main text provides additional discussion and comparison of all three structures.

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